

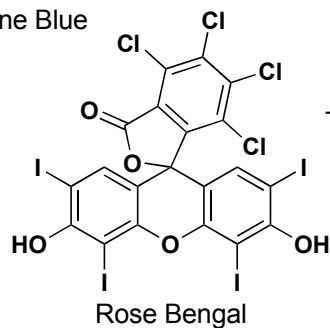
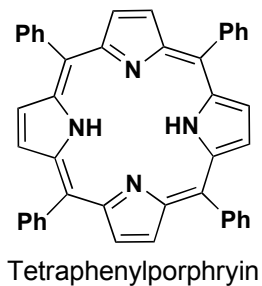
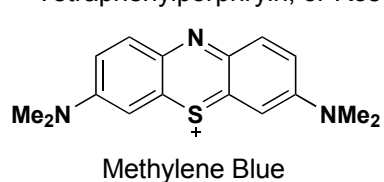
Characteristics of Singlet Oxygen

- 94.3 kJ/mol above Triplet State
- Singlet→Triplet transition is spin forbidden
- Lifetime of 74 minutes in gas phase
- Solution lifetime is highly solvent dependent
 - CCl₄ : 59 ms
 - Benzene : 30 μs
 - H₂O : 3.5 μs

Reviews: *Acc. Chem. Res.* **2008**, 1001
Tetrahedron, **2006**, 5308
Tetrahedron, **2005**, 6665
Pure Appl. Chem., **1975**, 481
Chem. Rev., **1981**, 91

Generation and Reaction Conditions

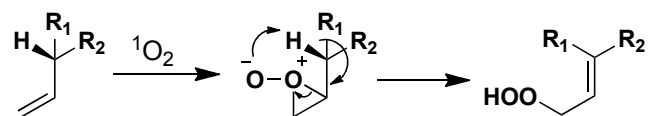
- Easily generated using light and a suitable photosensitizer
- Typically sensitized with Methylene Blue, Tetraphenylporphyrin, or Rose Bengal



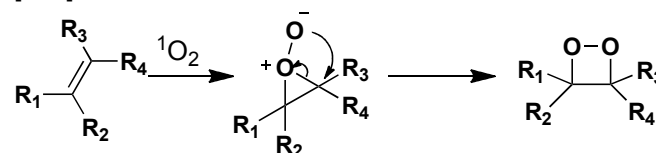
- Can also be generated by decomposition of hydrogen peroxide or organic peroxides
- Typical reaction conditions involve irradiation at low temperature because of the instability of many organic peroxides

Prototypical Singlet Oxygen Reactions

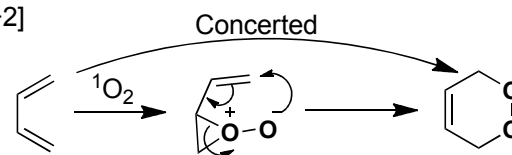
Ene:



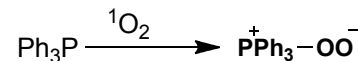
[2+2]



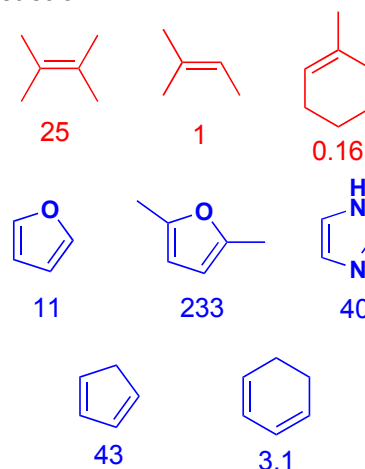
[4+2]



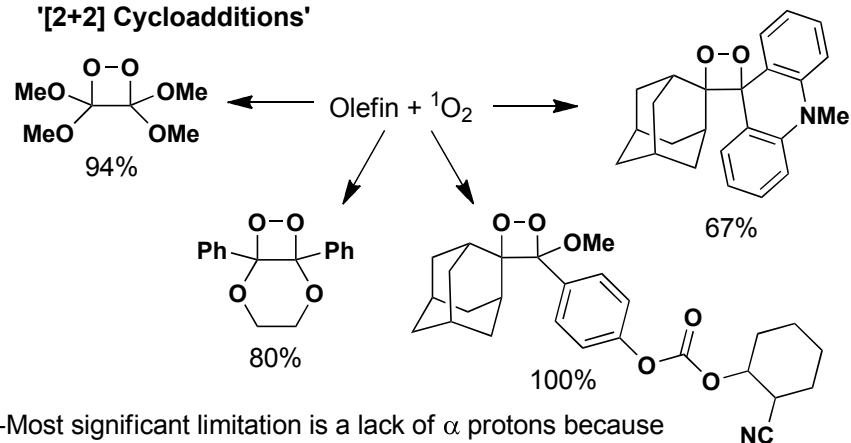
Heteroatom Oxidation

*Chem. Rev.*, **2008**, 1052

Relative Rates of Reaction

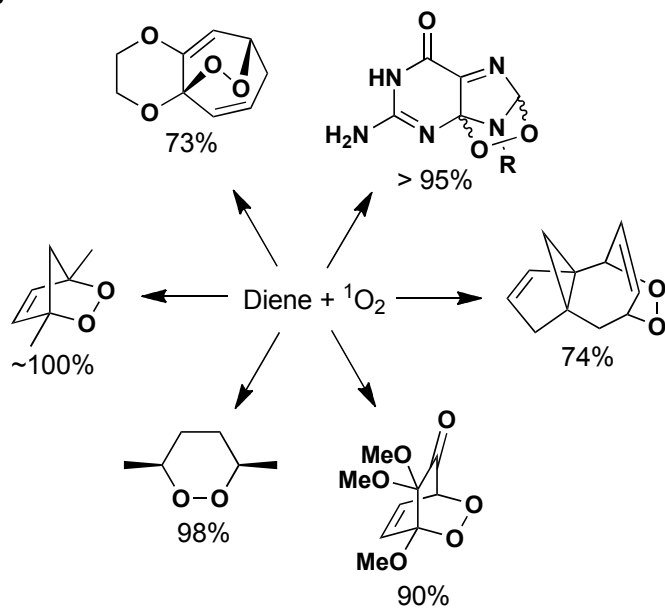
Frimer, A. A., *Singlet O₂*, **1985**

'[2+2] Cycloadditions'



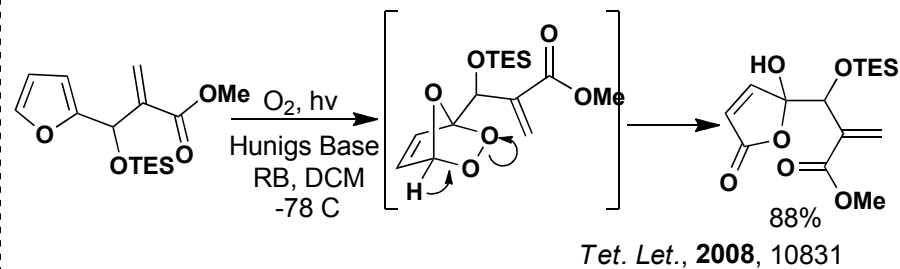
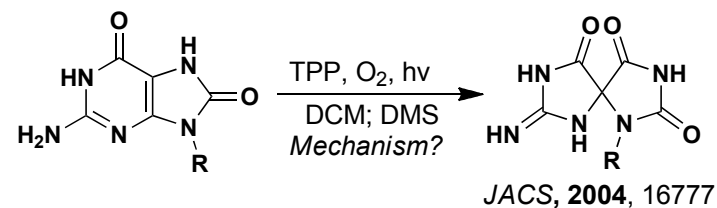
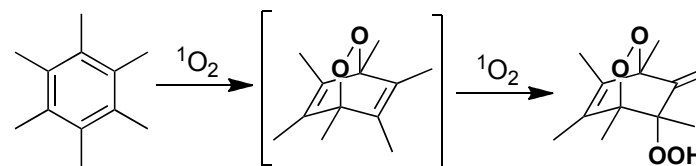
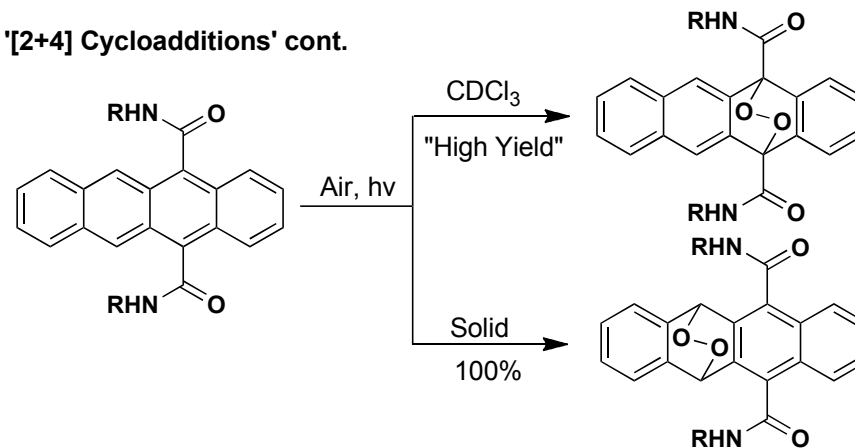
-Most significant limitation is a lack of α protons because the competing 'ene' reaction is significantly faster.

'[2+4] Cycloadditions'

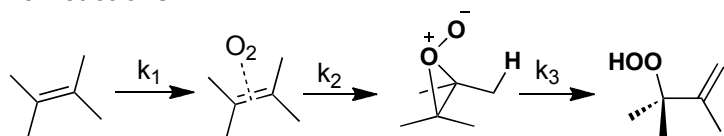


Handbook of Synthetic Photochemistry, 353

'[2+4] Cycloadditions' cont.



Ene Reactions



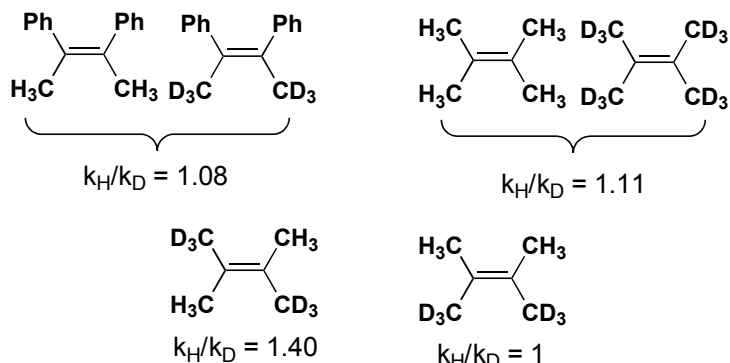
Mechanistic Details:

Exciplex Formation:

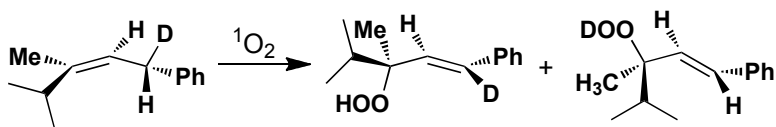
- Rate of reaction is $> 10^3$ less than diffusion control
- Reactions are characterized by low (or negative) activation enthalpies and high activation entropy

Periperoxide:

- Inter vs Intramolecular Isotope Effects



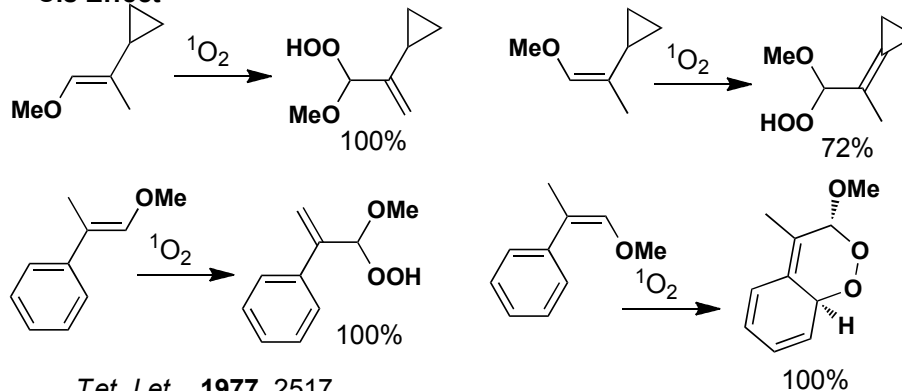
Ene Reaction is Suprafacial



Tetrahedron, 2000, 9151

Regioselectivity of the Ene Reaction

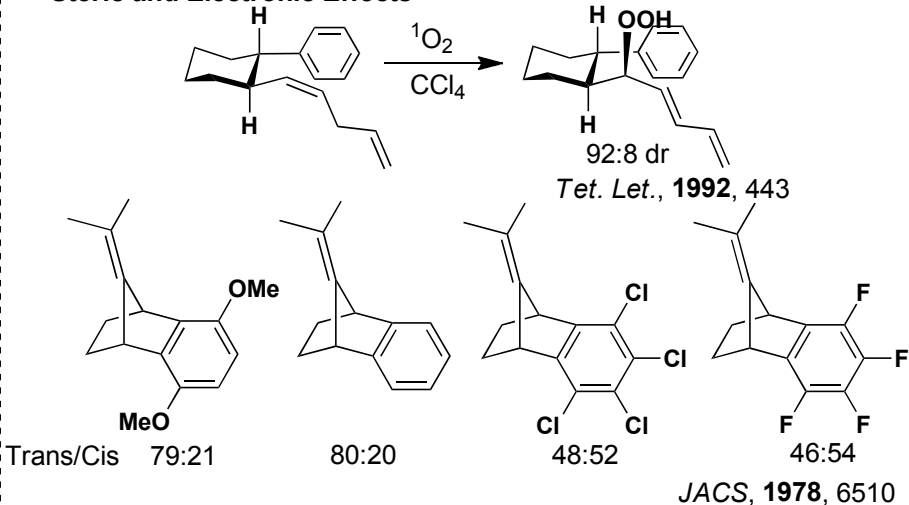
"Cis Effect"



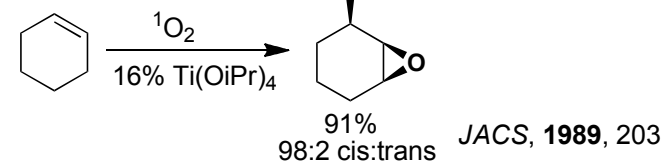
Tet. Let., 1977, 2517

Tet. Let., 1978, 3227

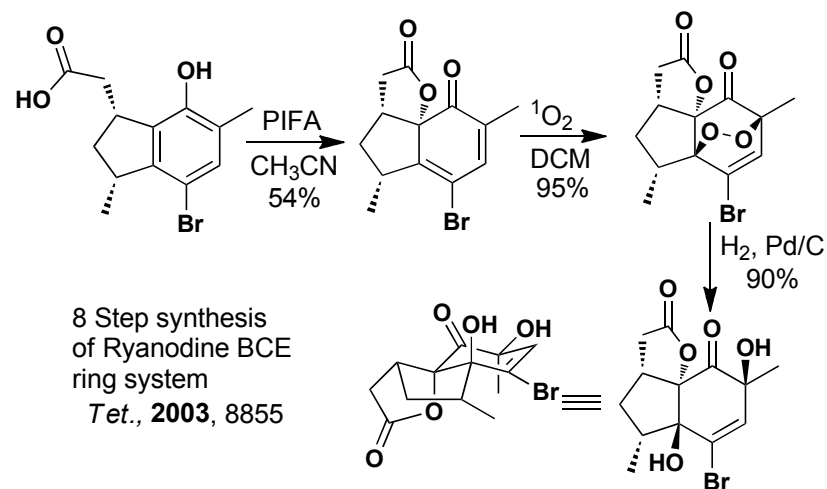
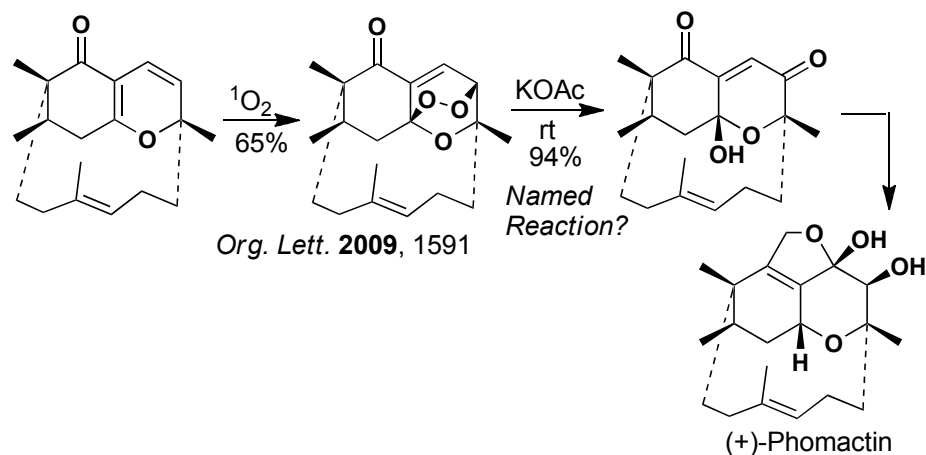
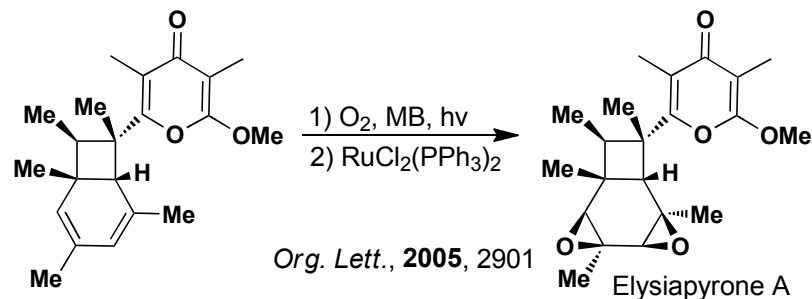
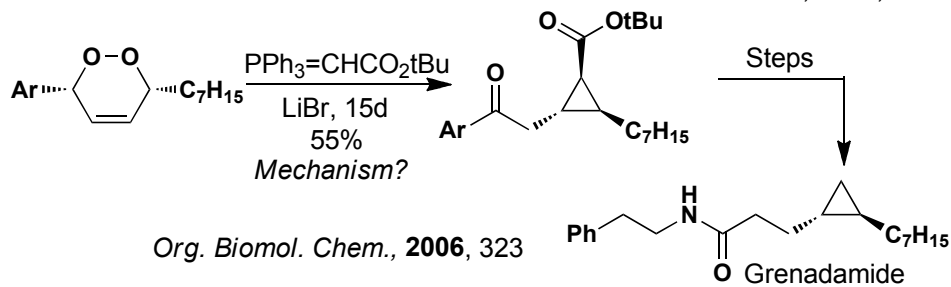
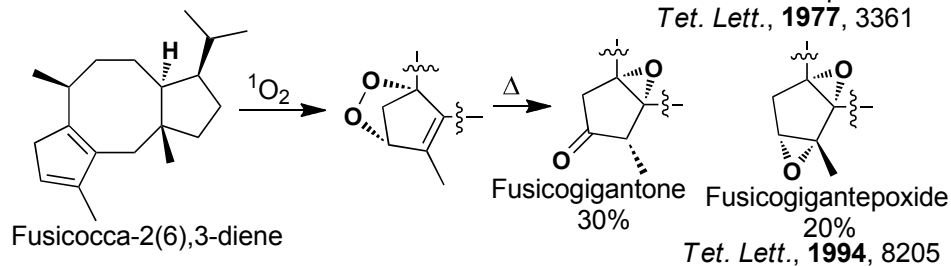
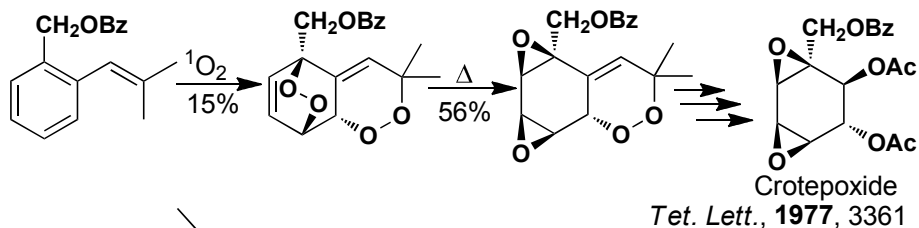
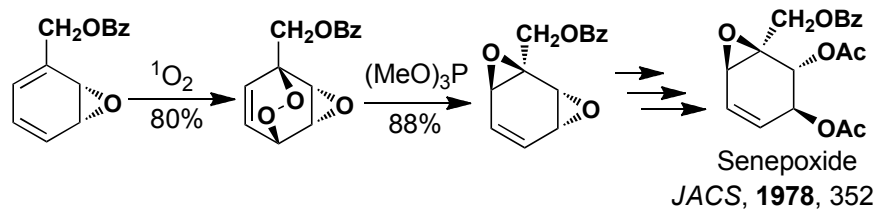
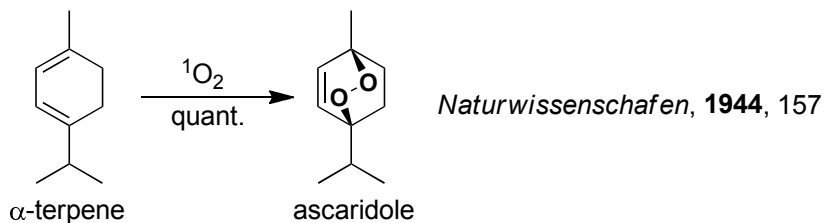
Steric and Electronic Effects



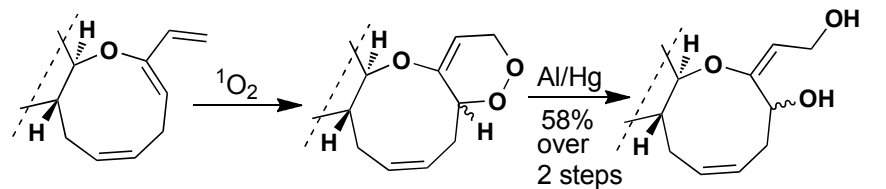
In-situ ene/epoxidation



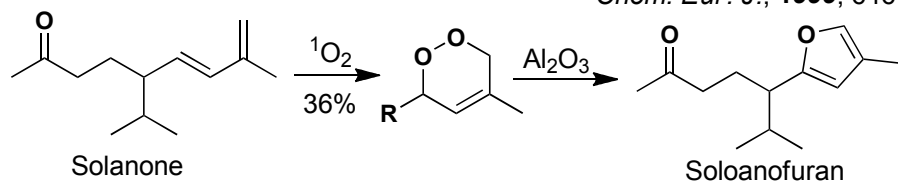
Synthetic applications of [4+2] and of resulting endoperoxides



Synthetic applications of [4+2] and of resulting endoperoxides (cont).

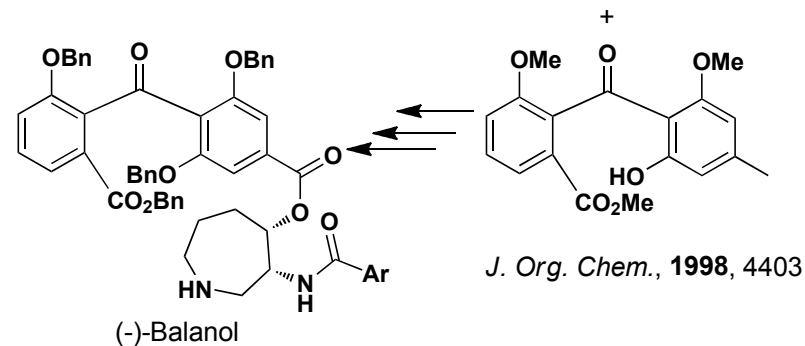
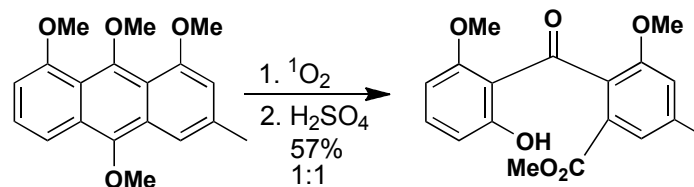
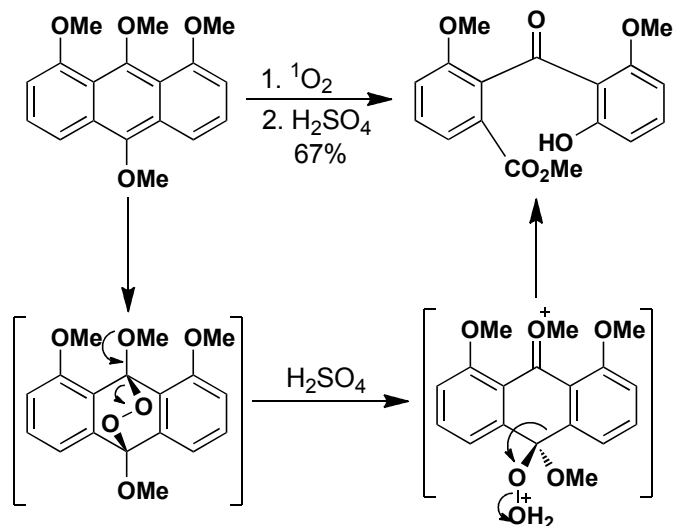
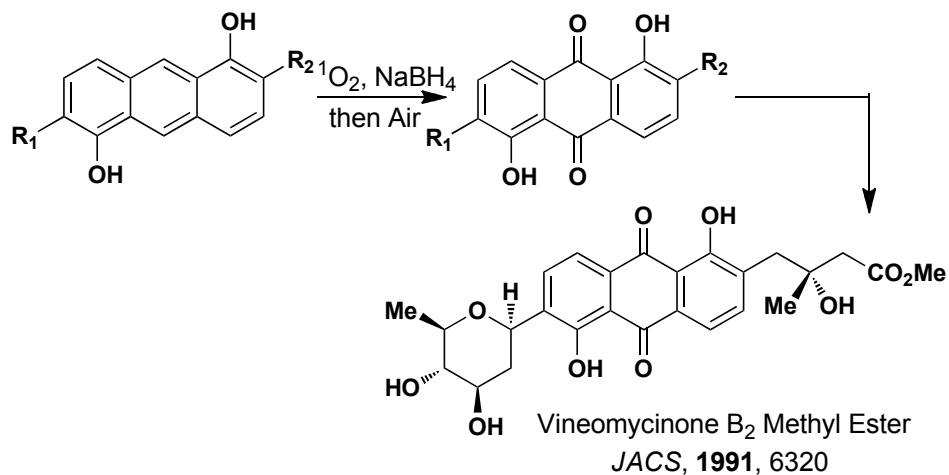
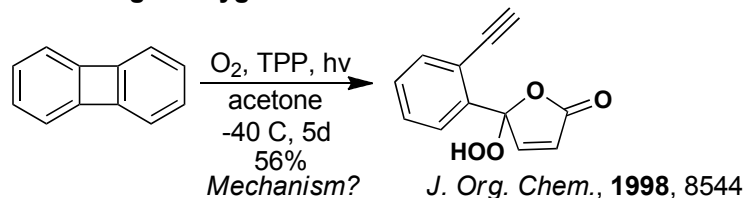


Chem. Eur. J., **1999**, 646

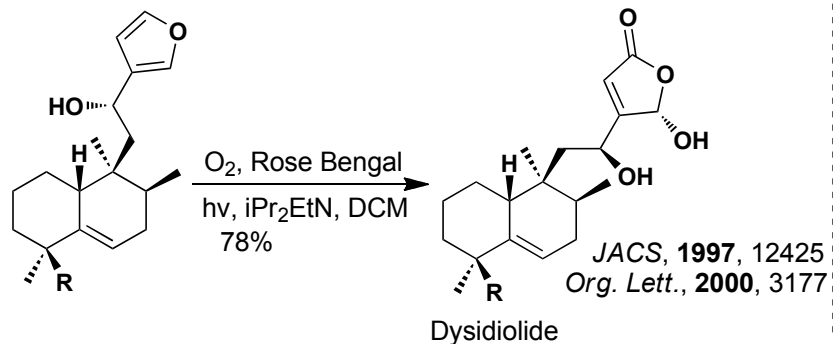
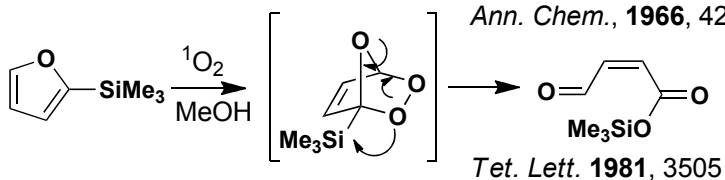
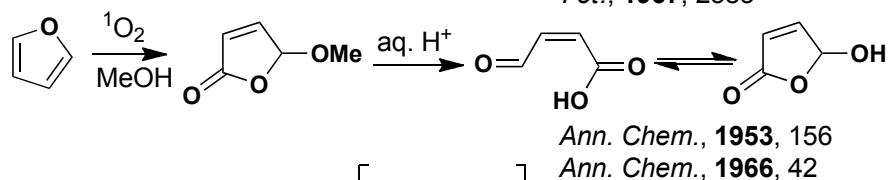
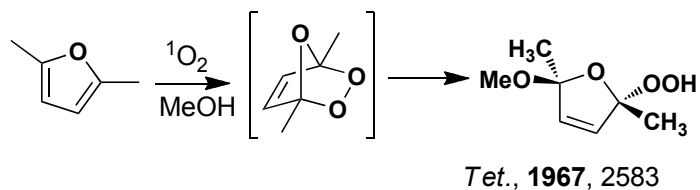
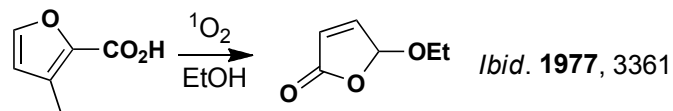
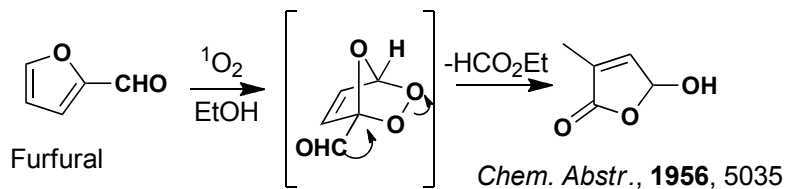


Helv. Chem. Act., **1972**, 265

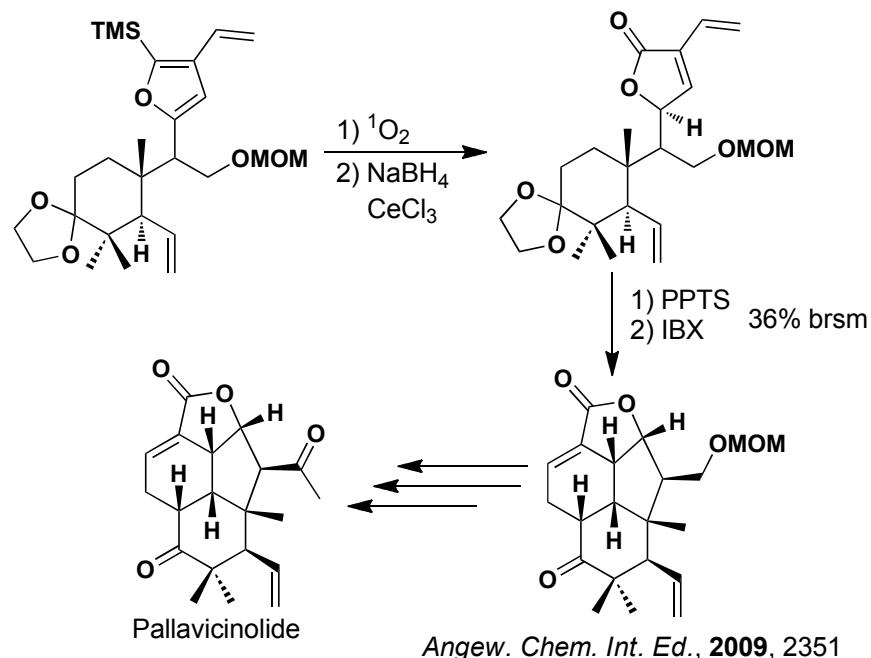
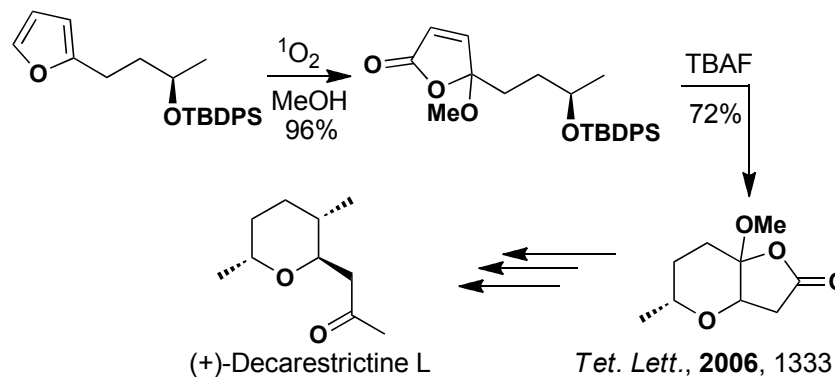
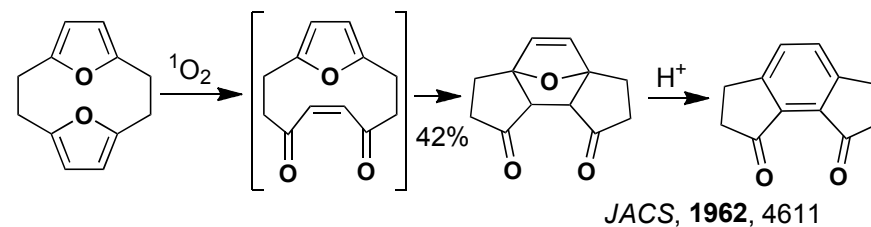
Reactions of Singlet Oxygen with Arenes



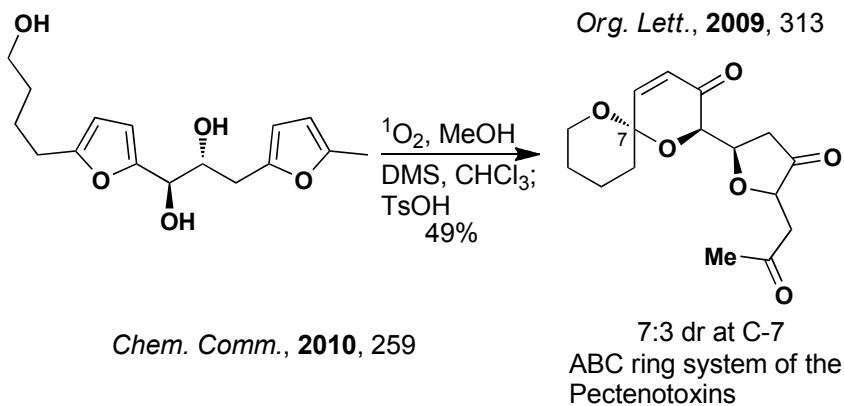
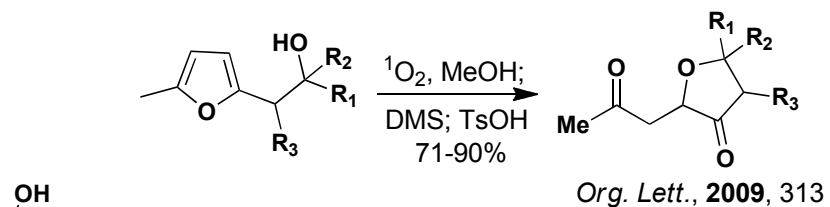
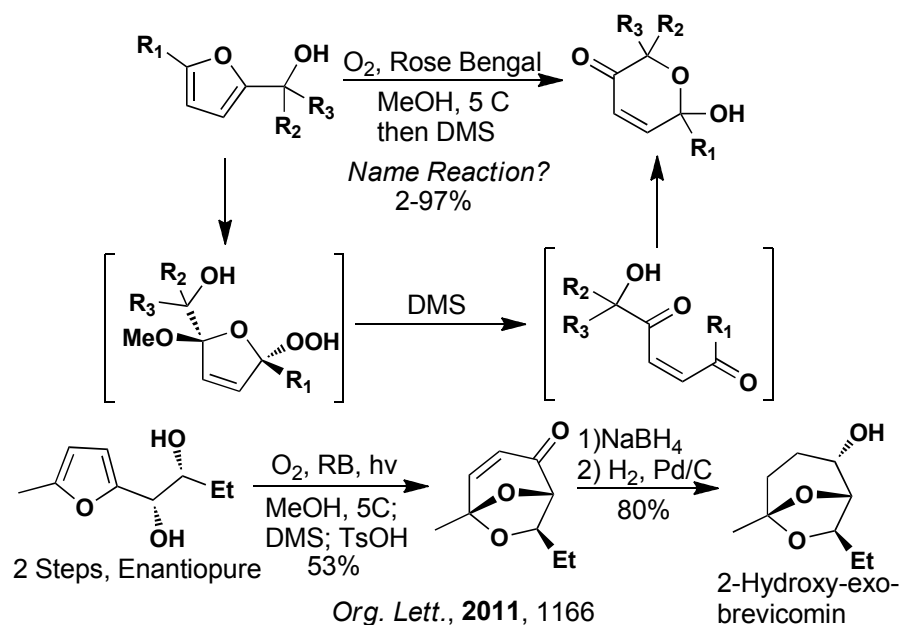
Reactions of Singlet Oxygen with Furans



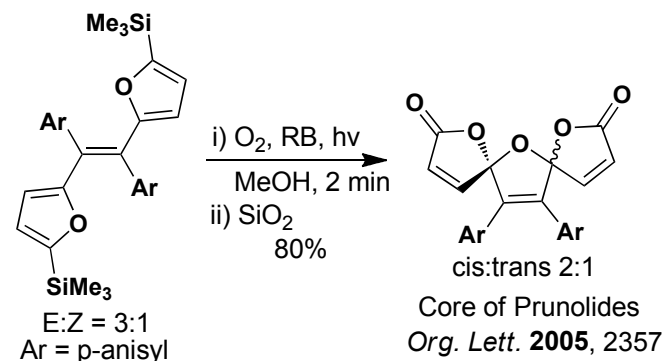
Dysidiolide



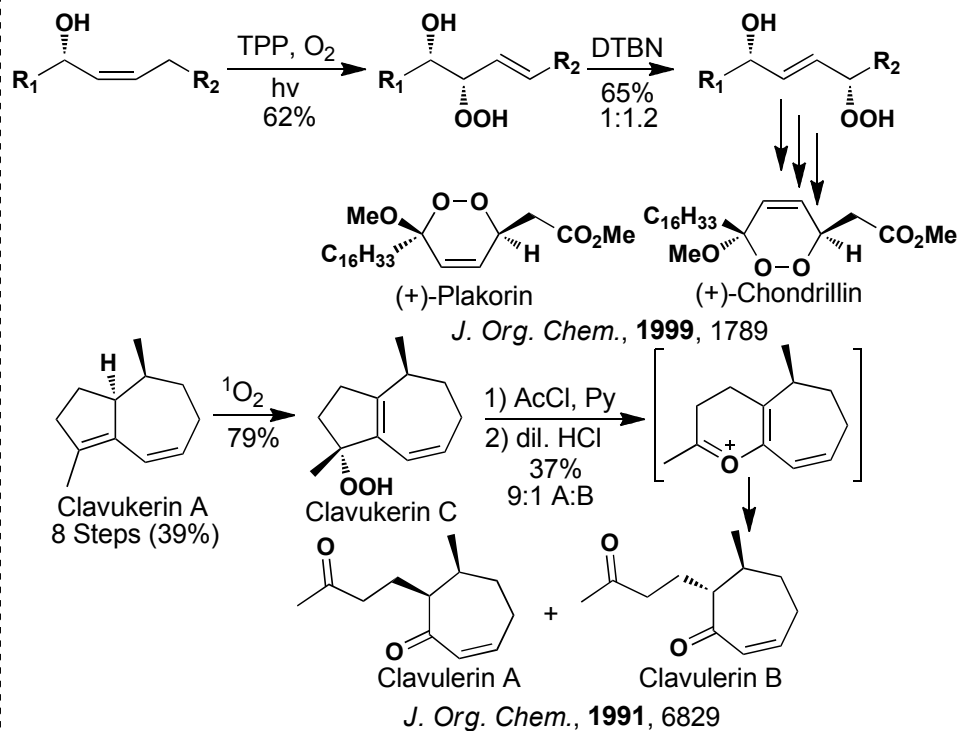
Using Furan Endoperoxides in Cascade Reactions



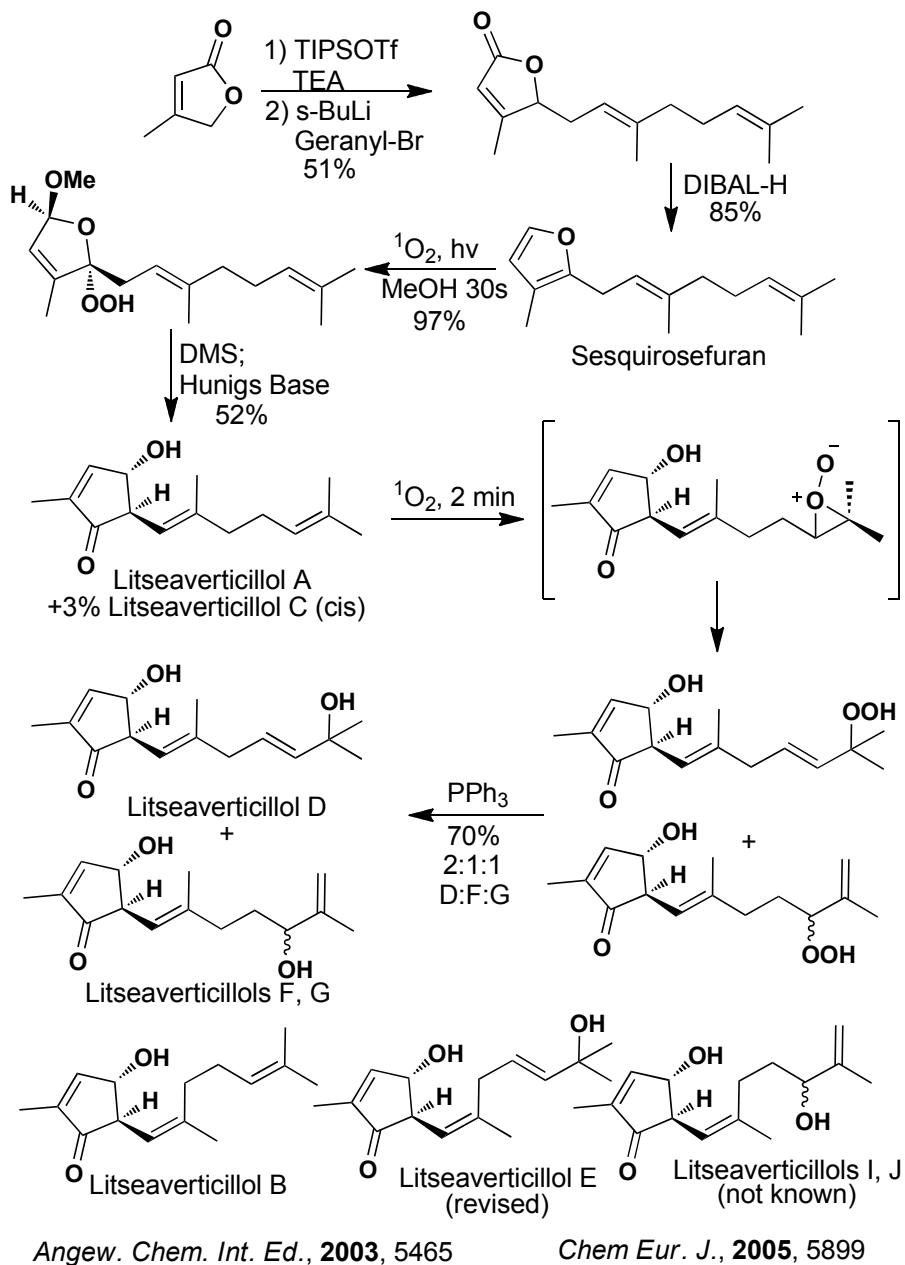
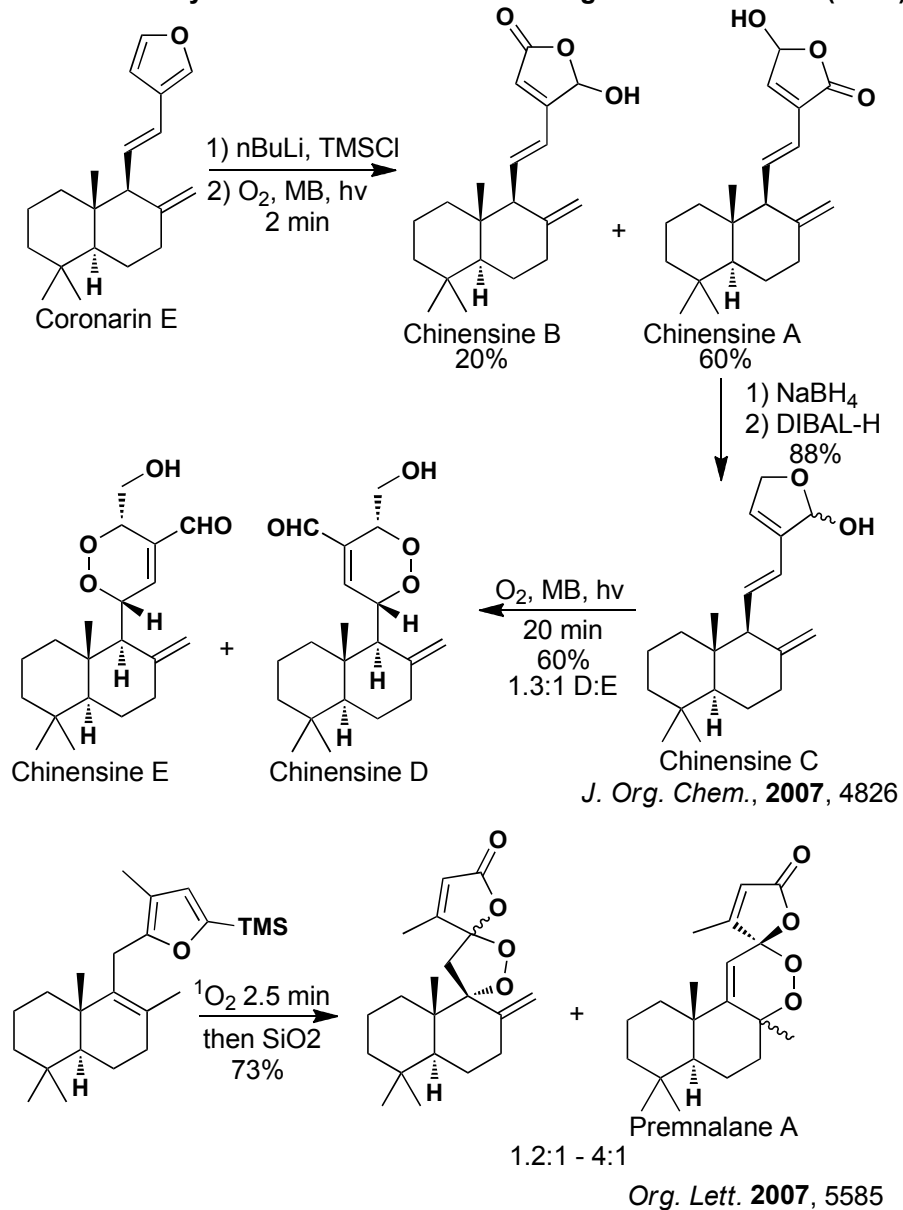
Using Furan Endoperoxides in Cascade Reactions (cont.)



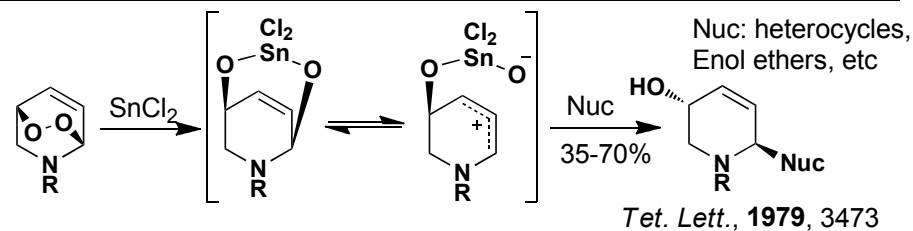
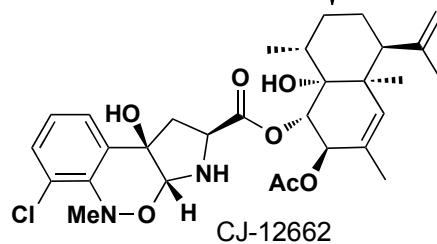
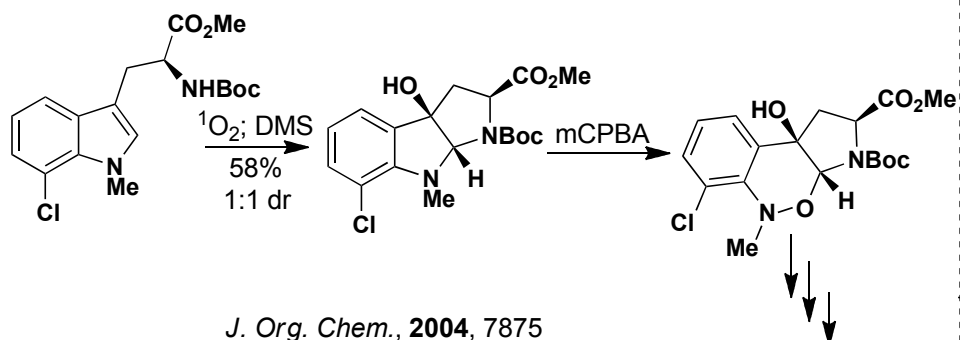
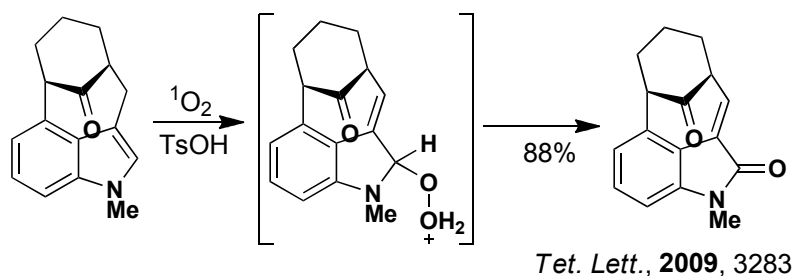
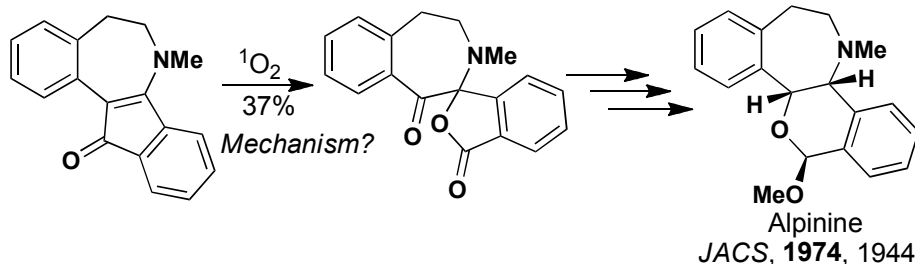
Biomimetic Synthesis of Peroxide Containing Natural Products



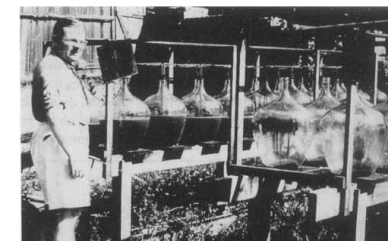
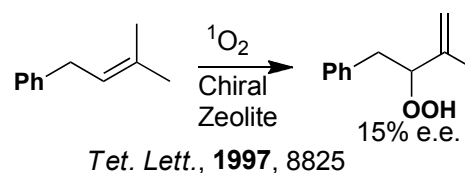
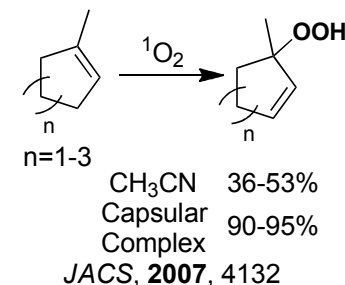
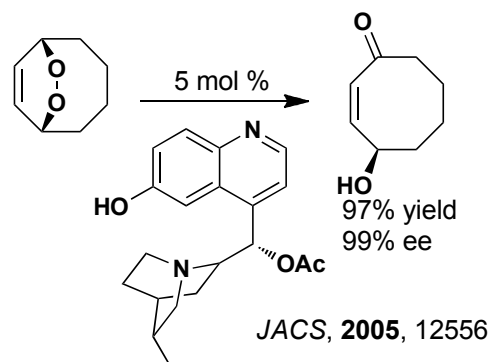
Biomimetic Synthesis of Peroxide Containing Natural Products (cont.)



Reactions of other heterocycles with Singlet Oxygen



Misc. Singlet Oxygen Chemistry

Schenck at his ascaridol pilot plant in 1952
Solar Photochemistry: *Angew. Chem. Int. Ed.*, 1994, 2009

Other Applications of Singlet Oxygen Not Covered

