

## Disclaimer

This group meeting is in no way meant to be comprehensive. There is an enormous body of literature featuring sugar starting materials in total synthesis of which this group meeting is only a sample. Any comparison made between syntheses are for educational and discussion purposes only.

## Resources

The following books and reviews are excellent resources and were very helpful in the preparation of this group meeting.

Total Synthesis of Natural Products:  
The "Chiron Approach"  
Stephen Hanessian ISBN-10: 0080307159

Design and Strategy in Organic Synthesis  
Stephen Hanessian ISBN-10: 3527319646

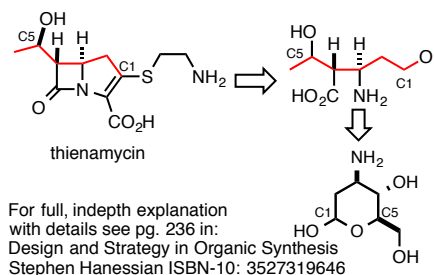
Organic Synthesis with Carbohydrates  
Geert-Jan Boons ISBN-10: 1850759138

		World production <sup>a</sup> (metric t/year)	Price <sup>b</sup> (€/kg)
Sugars	Sucrose	130.000.000	0.30
	D-Glucose	5.000.000	0.60
	Lactose	295.000	0.60
	D-Fructose	60.000	1.00
	Isomaltulose	50.000	2.00
	Maltose	3.000	3.00
	D-Xylose	25.000	4.50
Sugar Alcohols	L-Sorbose	60.000	7.50
	D-Sorbitol	650.000	1.80
	D-Xylitol	30.000	5.00
	D-Mannitol	30.000	8.00
Sugar-derived Acids	D-Gluconic aci	60.000	1.40
	L-Lactic aci	> 100.000	1.75
	Citric aci	500.000	2.50
	L-Tartaric acid	35.000	6.00
Amino Acids	L-Lysine	40.000	5.50
	L-Glutamic acid	500.000	7.00

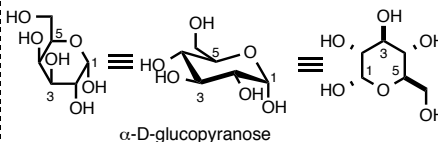
## Tracking Treasure Down: Hanessian's "Rule of Five"

- 1) Select an sp, sp<sup>2</sup>, or sp<sup>3</sup>-hybridized carbon atom in target structure
- 2) Move five bonds away and look for a heteroatom regardless of functionality in between. These two internal reference points correspond to the anomeric C-1 and the C-5 oxygen atoms in the hexapyranose structure
- 3) Look at only that carbon chain and try to imagine a suitable chiral sugar SM.

## An Example:



## Different Representations

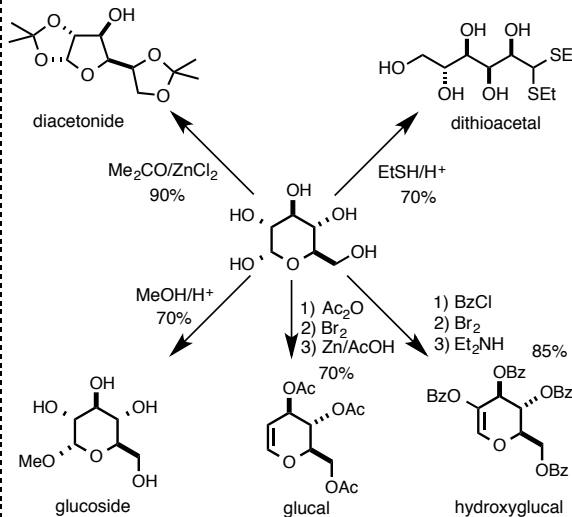


## Molecular Target Categories

The target structure can be broken down into three categories based on how easily it can be traced back to the chiral carbohydrate pool.

## Typical Reaction Channels

Working from a feedstock carbohydrate is difficult, time consuming and wasteful. Entry into any reaction channel toward an enantiopure building block is mostly limited to chemistry dating back over 100 years in the carbohydrate literature. The initial stage always involves the fixation of the sugar in the respective tautomeric form.



## Cat. 1: Apparent Carbohydrates

Usually a lightly modified carbohydrate appended to a structure via glycosidation. Very easy to spot.

## Cat 2: Partially hidden carbohydrates

One or two chiral centers may have been removed. Still contains carb. oxidation pattern. Identification likely requires squinting.

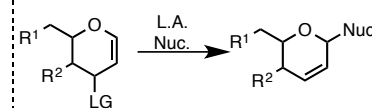
## Cat. 3: Totally Hidden Carbohydrates

Identification of carb. SM unlikely. Usually based around application of in-house methodology.

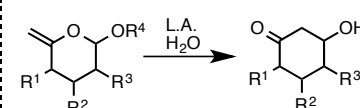
\*These are my simplified definitions based on Hanessian's own in depth definitions, examples and analysis.

## Important Reactions to Keep In Mind

### Type I Ferrier reaction



### Type II Ferrier rearrangement



## Natural Abundance

Carbohydrates are the single most abundant class of organic compounds associated with living matter.

ACS Symposium Series 841 2003, 47.

## Is sugar derived chirality right for me?

- End differentiated oxidation state
- Stereochemical diversity
- Carbon chain length varying between 3-7 carbons
- Strong conformational bias
- Ability to introduce other heteroatoms (N, S, X) with inversion or retention of configuration
- Ability to branch using a C nucleophile
- Excellent conformational bias and stereoelectronic effects maximize predictive outcomes based on models

## Why don't we see more syntheses based on chiral sugars?

- Overfunctionalization with hydroxyl groups that have similar or identical reactivity
- Number of chiral centers present on starting carbohydrate often excessive for the synthetic chemist
- Lack of suitable functional groups (olefin, carbonyl) to which modern organic techniques can be applied
- Outcome of chemical transformations are unique to each sugar based on individual stereochemistry
- Development of outstanding asymmetric methodology

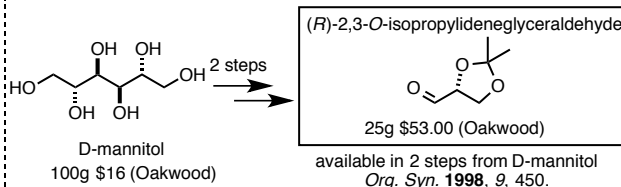
## Full exploitation of sugar starting materials

- Retention of the carbon-chain of the sugar
- Reaction sequence should be simple, high yielding with no chromatography. Focus on generating crystalline products.
- Avoid protecting groups or use simple, easy to remove PGs
- Cost.

## On practicality: reality vs. raw concept

There are a huge variety of chiral carbohydrates available in principle. However, the pool shrinks immensely when practical and operational considerations are applied leaving a relatively small number of realistic chiral sugar starting materials.

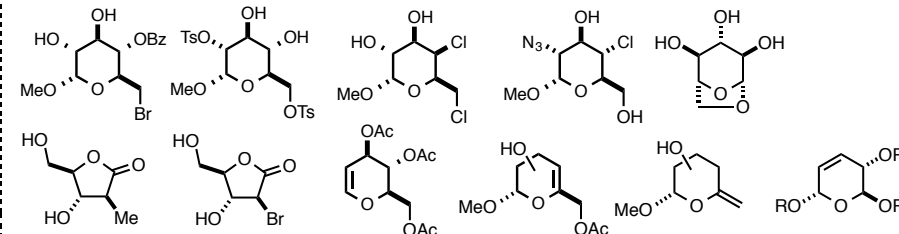
## Arguably the Most Useful Sugar Based "Chiron" for Modern Total Synthesis



For every one unit of D-mannitol subjected to protection/oxidative cleavage two desired aldehydes are formed. The resulting aldehyde is highly prone to polymerization and hydrolysis. Long term storage is not recommended

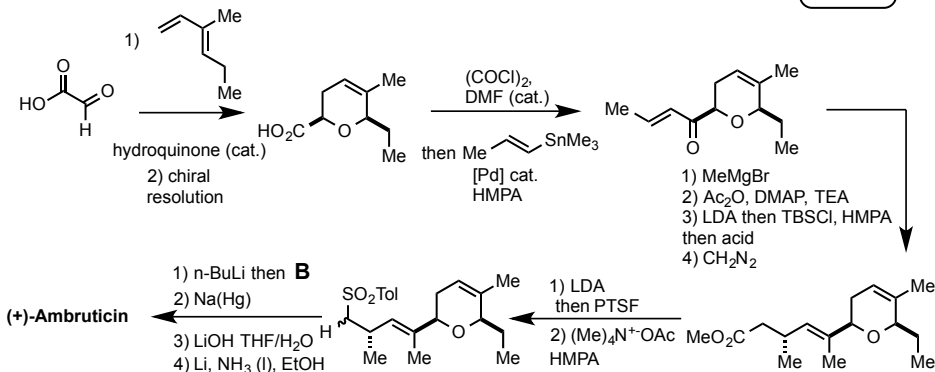
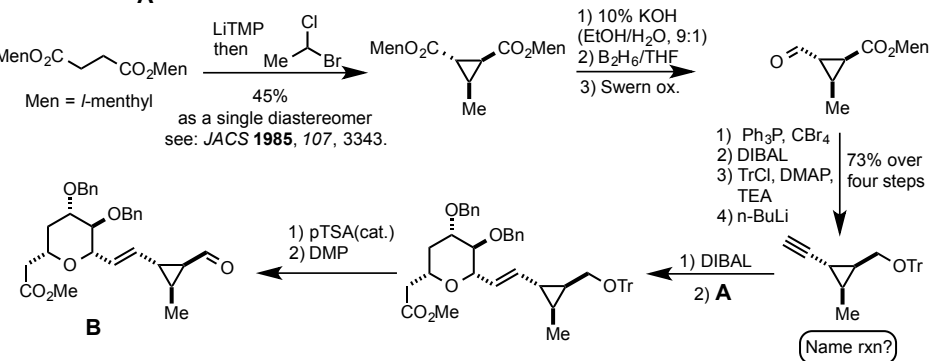
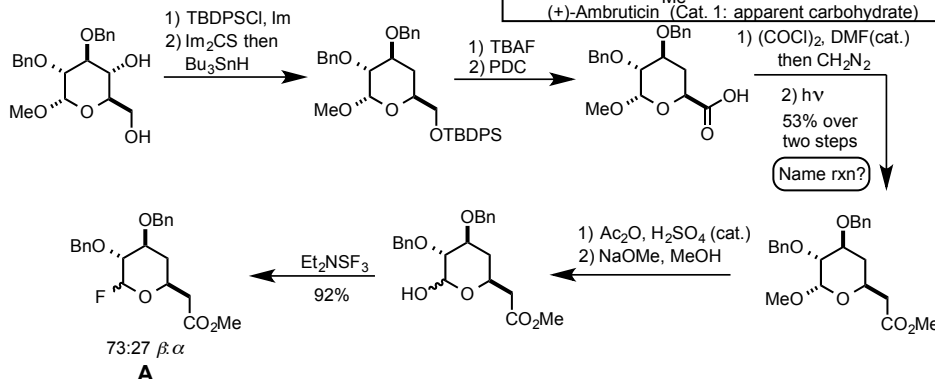
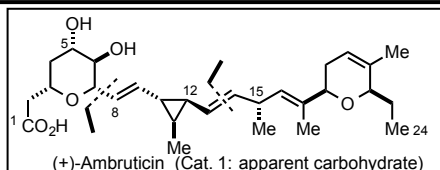
For (S)-derivative see:  
Org. Syn. 1995, 72, 1.  
Synlett 2001, 10, 1565.

## Some Easily Accessible/Commercial/Useful Sugar Based Chirons



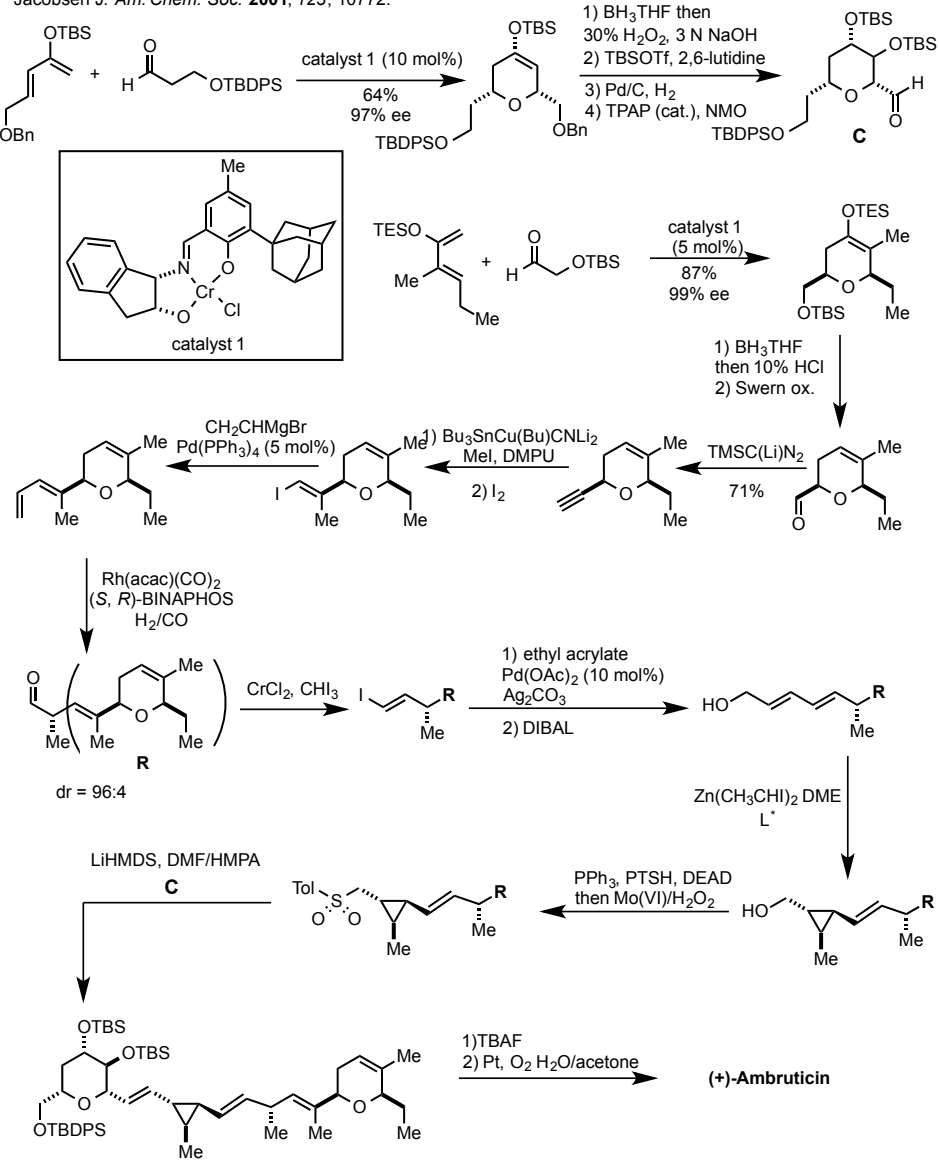
## Total Synthesis of (+)-Ambruticin Kende's chiral sugar approach

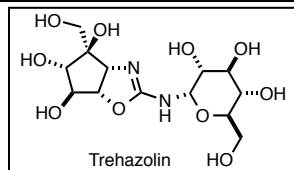
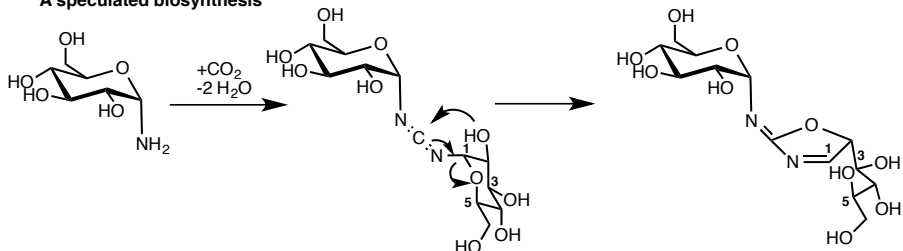
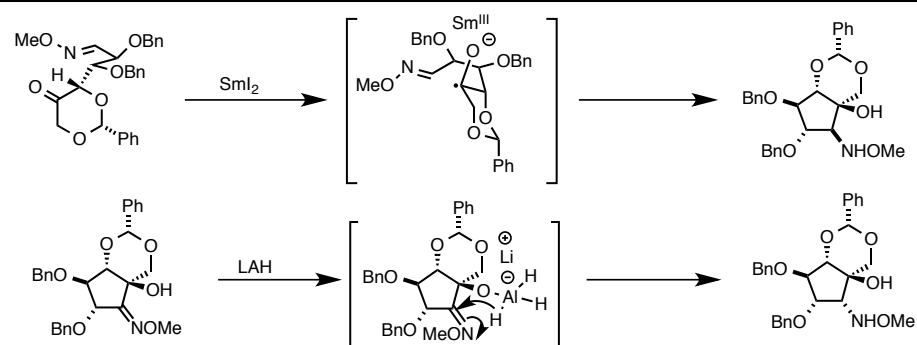
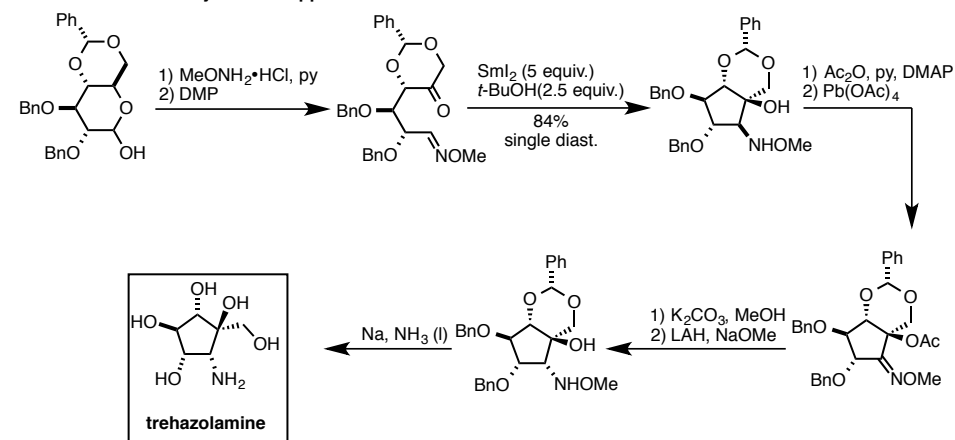
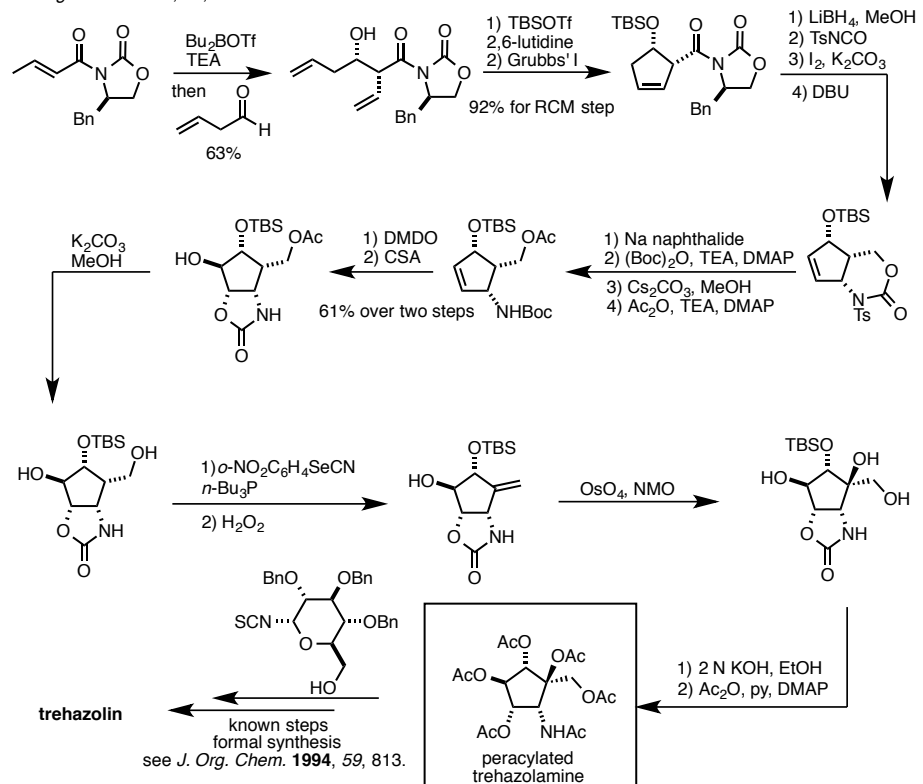
Kende, A. J. *Am. Chem. Soc.* **1990**, *112*, 9645.



## Total Synthesis of (+)-Ambruticin Jacobsen's asymmetric catalysis approach

Jacobsen J. *Am. Chem. Soc.* **2001**, *123*, 10772.

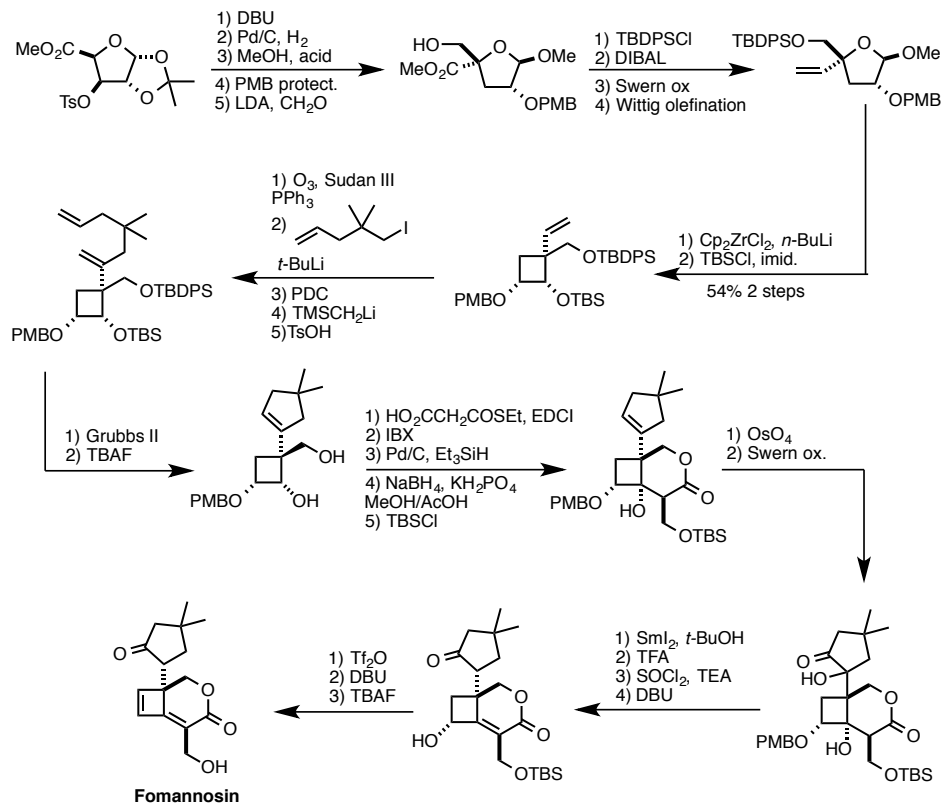


**Synthesis of Trehazolin from D-Glucose: following a plausible biosynthesis**Giese, B. *J. Org. Chem.* **1998**, *63*, 5877.Trehazolin from trehazolamine is known in the literature see *J. Org. Chem.* **1994**, *59*, 813.**A speculated biosynthesis****Giese's reductive cyclization approach****The Crimmins' Synthesis of Trehazolin: The chiral auxiliary and RCM approach***J. Org. Chem.* **2001**, *66*, 4012.



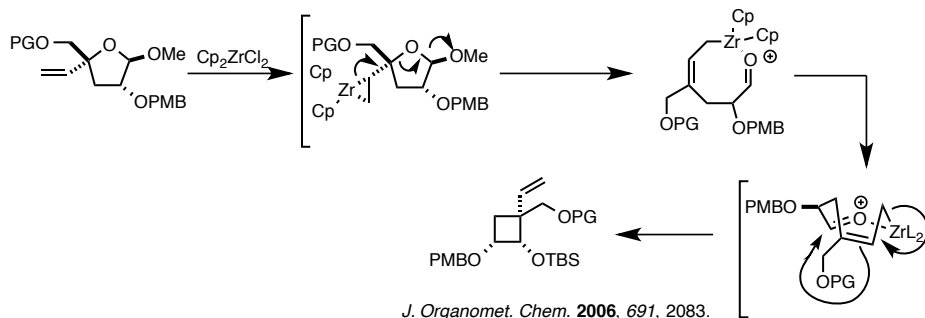
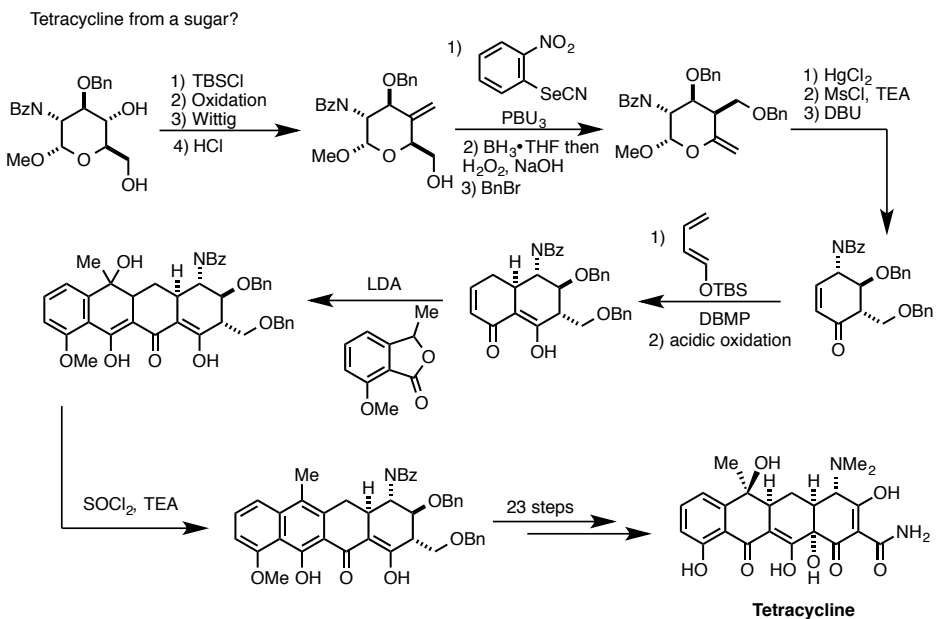
## Total Synthesis of Fomannosin

Paquette, L. *Angew. Chem. Int. Ed.* **2007**, *46*, 7817.



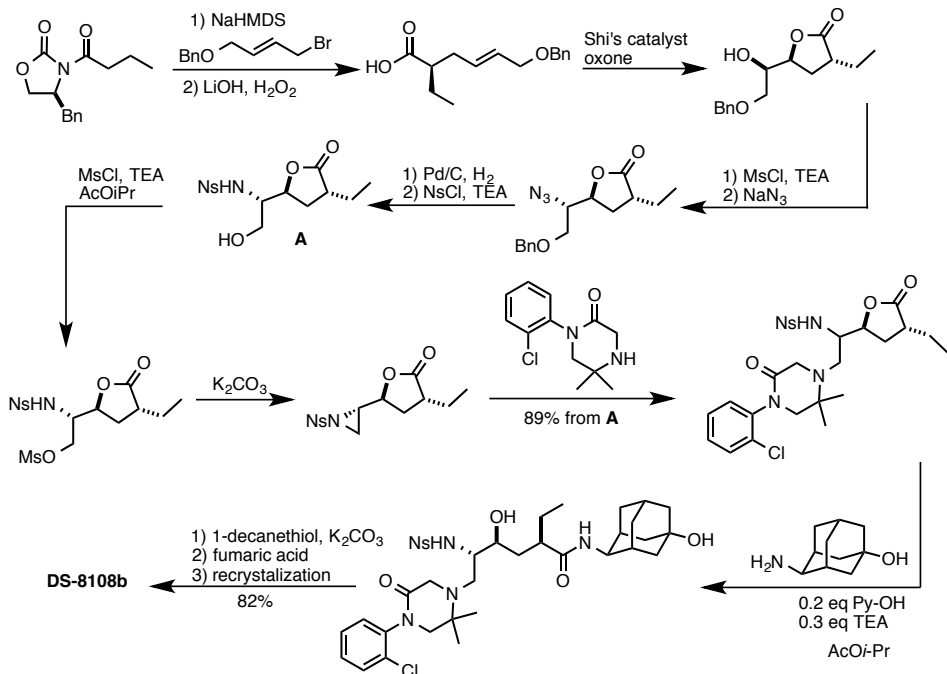
## Total Synthesis of (-)-Tetracycline

Tatsuta, K. *Chem. Lett.* **2000**, 647.

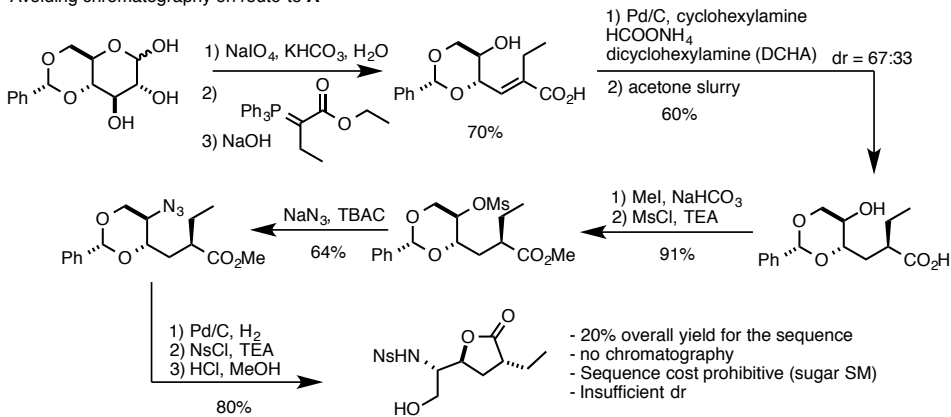


**Synthesis of DS-8108b: from chiral aux to sugar and back**  
Daiichi Sankyo Co. *Org. Process Res. Dev.* 2013, 17, 1430.

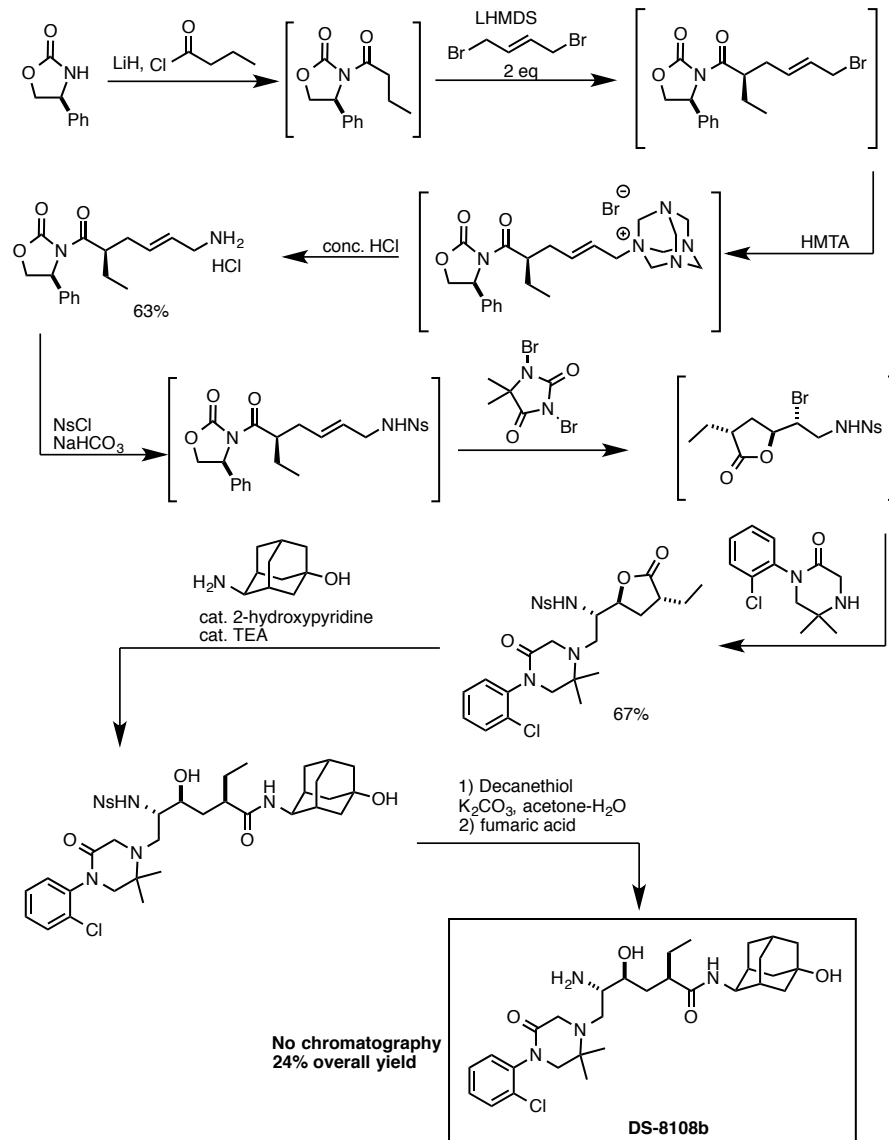
**Early-stage synthesis**



**So where does the sugar come in?**  
Avoiding chromatography en route to A

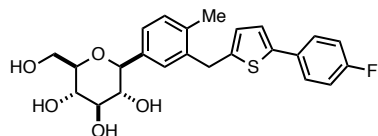


**DS-8108b continued**  
Sugars: just a bridesmaid, back to the chiral auxiliary  
Can your sugar do this?

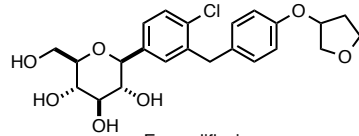


## Sugars: What are they good for?

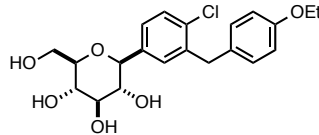
SGLT2 inhibitors. Sodium-glucose co-transporter 2 (SGLT2) inhibitors are a new class of drugs for the treatment of type 2 diabetes. Type 2 diabetes affects 23 million Americans making this area particularly interesting for the development of novel treatments.



Canagliflozin  
FDA approved



Empagliflozin  
limited European approval



Dapagliflozin  
limited European approval

## SGLT2 inhibitors, how do they work?

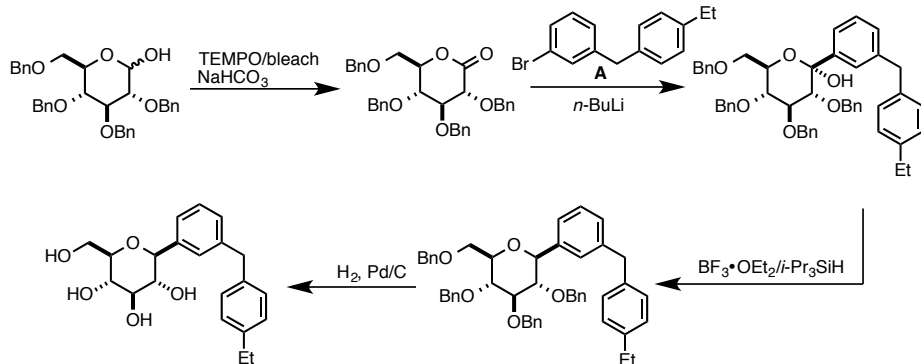
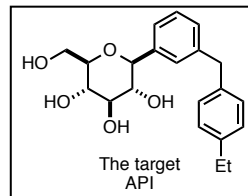
The SGLT family includes the Na/glucose co-transporter SGLT1, which is mainly expressed in the GI tract and is responsible for glucose absorption from the food intake. SGLT2 is mainly expressed in the kidneys. SGLT2 is responsible for ~90% of glucose reabsorption in humans. SGLT2 inhibitors work by blocking the reabsorption of glucose in the kidneys. This prevents the build up of glucose in the blood and allows for excess glucose to be eliminated via urination. One of the most common side effects is genital infection.

For an indepth discussion see: *Current Pharmaceutical Design* **2014**, *20*, 3647.

## Development of a large scale synthesis of an SGLT-2 inhibitor

Bristol-Myers Squibb *OPRD* **2012**, *16*, 577.

First-generation synthesis



## Process R & D synthetic route: a tale of two protecting groups

