

Cyclopropanes in Synthesis

J. Gallagher

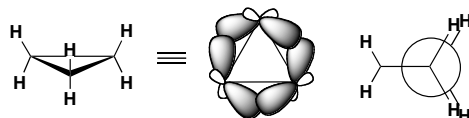
Wednesday, February 18, 2004

An Brief History of Cyclopropanes-

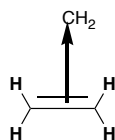
- 1884 First synthesis of a cyclopropane derivative by William Henry Perkin with Adolf von Baeyer's
- 1896 Vinylcyclopropane synthesized by Gustavson.
- 1922 Existence of vinylcyclopropane secured, yet faces despite residual disbelief.
- 1931 Pauling describes the banana bonding model of cyclopropane
- 1942 Cyclopropanol synthesized by Cottle.
- 1947 Walsh describes cyclopropane as a coordinate bond between ethylene and :CH_2
- 1952 Hexafluorocyclopropane generated by TFE photolysis.
- 1954 Doering et al. find dichlorocarbene adds into alkenes to generate *gem*-dichlorocyclopropane.
- 1957 Triphenyl cyclopropenyl cation synthesized by Breslow, constituting the smallest aromatic species possible.
- 1958 Simmons and Smith determine that iodoalkylzinc species are capable of cyclopropanation.
- 1964 Wittig elaborates diazoalkanes as an alternative to iodoalkane precursors of organozinc species.

Physical Descriptions of Cyclopropane-

-Pauling's sp^3 , "banana" bonded cyclopropane: This model suggests that sp^3 hybridized carbons give rise to a stable, trigonal planar arrangement. Further instability is inferred as all hydrogens are eclipsed.

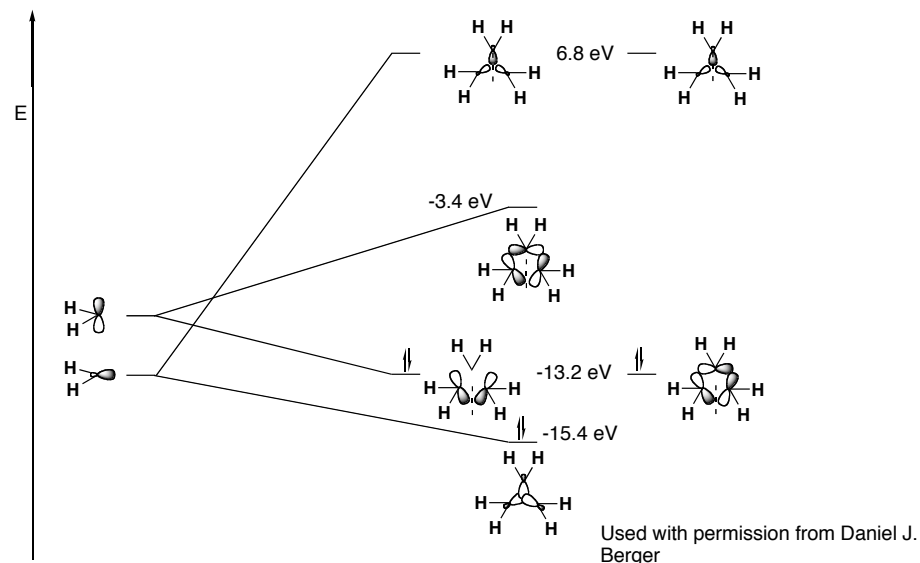


-Walsh cyclopropane description: In an analogy to epoxides, A.D. Walsh proposed that cyclopropanes be considered as an insertion of methylene into ethylene, ultimately giving rise to the D_{3h} symmetric product.



Thus, Walsh cyclopropane has significant sp^2 character and should react in analogy to olefins; further, as reactivity is generally governed by HOMO-LUMO chemistry, one can also see why Walsh cyclopropane has a propensity to open, yet while it is also a stable compound. The dative bond suggestion was immediately hated by the British chemical community, with Sir. R. Robinson leading the criticism. Walsh's suggestions had been picked apart due to his suggestion of subtle similarities between ethylene oxide/cyclopropane and N-oxide species, ethylene, and the relationship between chemical character and depiction. Experimental support, however, was presented a year later for cyclopropane's olefinic character, strengthening Walsh's hypothesis.

Walsh Orbital Energies



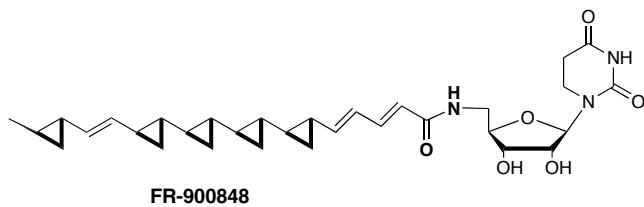
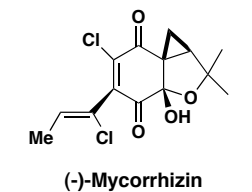
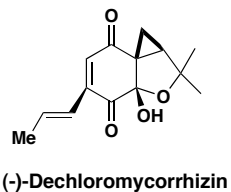
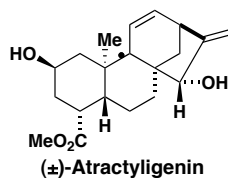
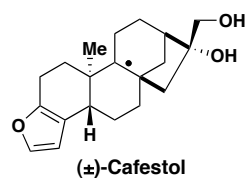
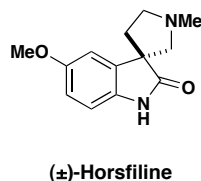
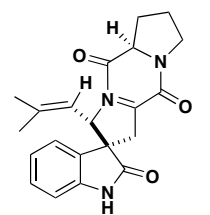
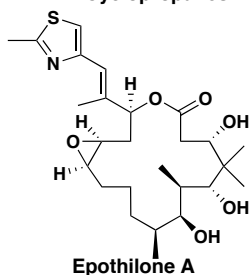
Outline of Cyclopropanes in Synthesis:

- **Stereocontrol via cyclopropanation:**
 - Corey's syntheses of (\pm)-Atractyligenin and (\pm)-Cafestol
 - Smith's synthesis of (\pm)-MycorrhizinA and (\pm)-Dechloromycorrhizin A
- **Cyclopropanes as equivalents or masked groups:**
 - Danishefsky's synthesis of Epothilone A
 - Carreira's synthesis of (-)-Spirotryptostatin and (\pm)-Horsfilin
- **Synthesis of unusual cyclopropanated natural products:**
 - Barrett's synthesis of FR-900848
 - Faick's synthesis of FR-900848

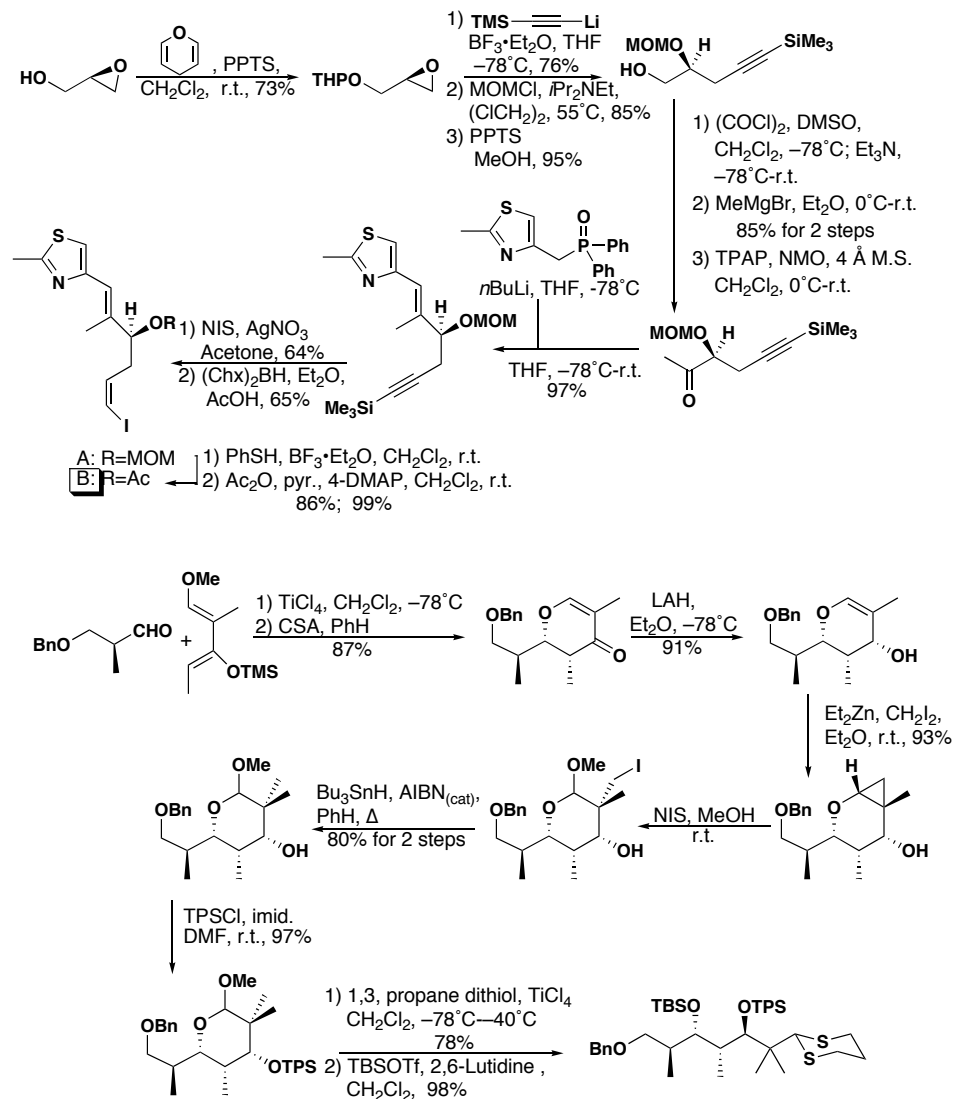
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Cyclopropanes in Selected Total Syntheses:



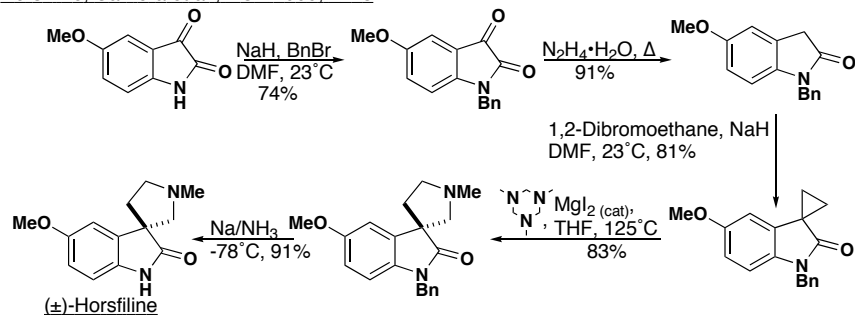
Epothilone A: Danishefsky et al., JACS, 1997, 10073



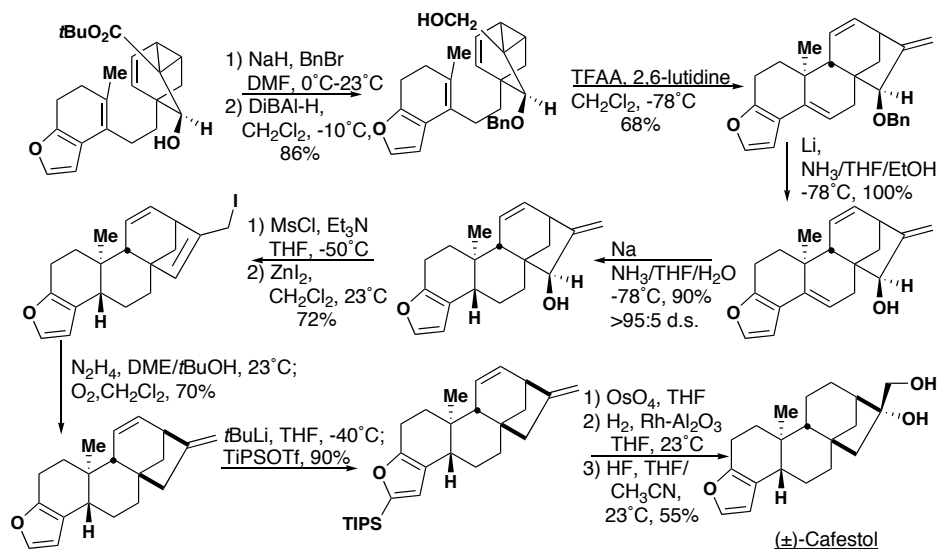
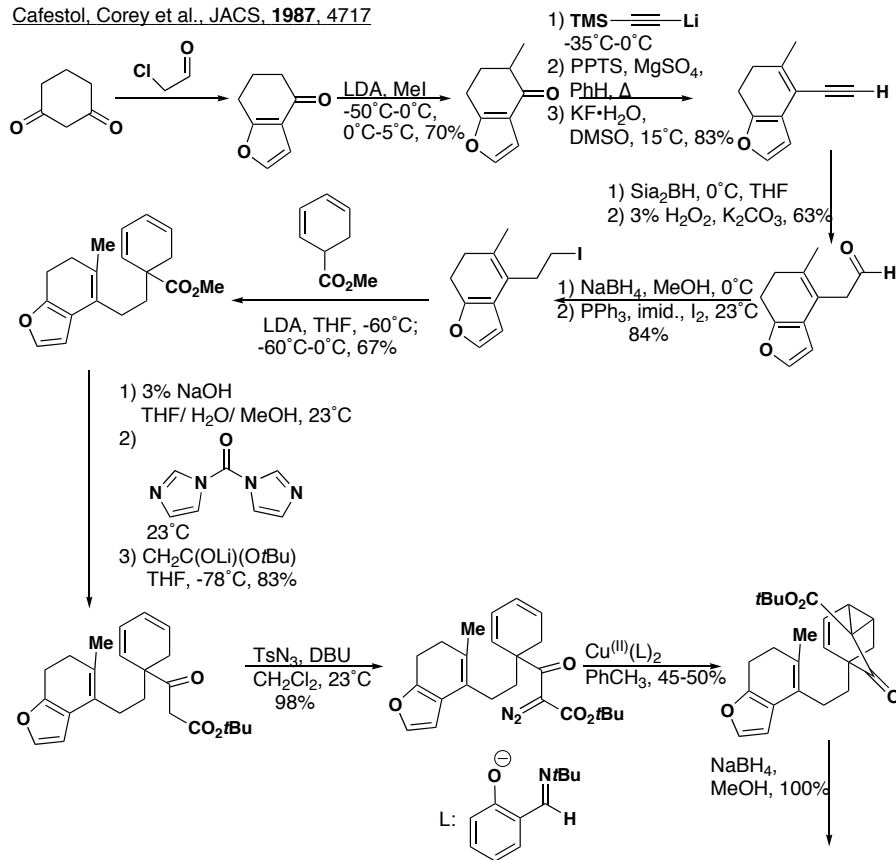
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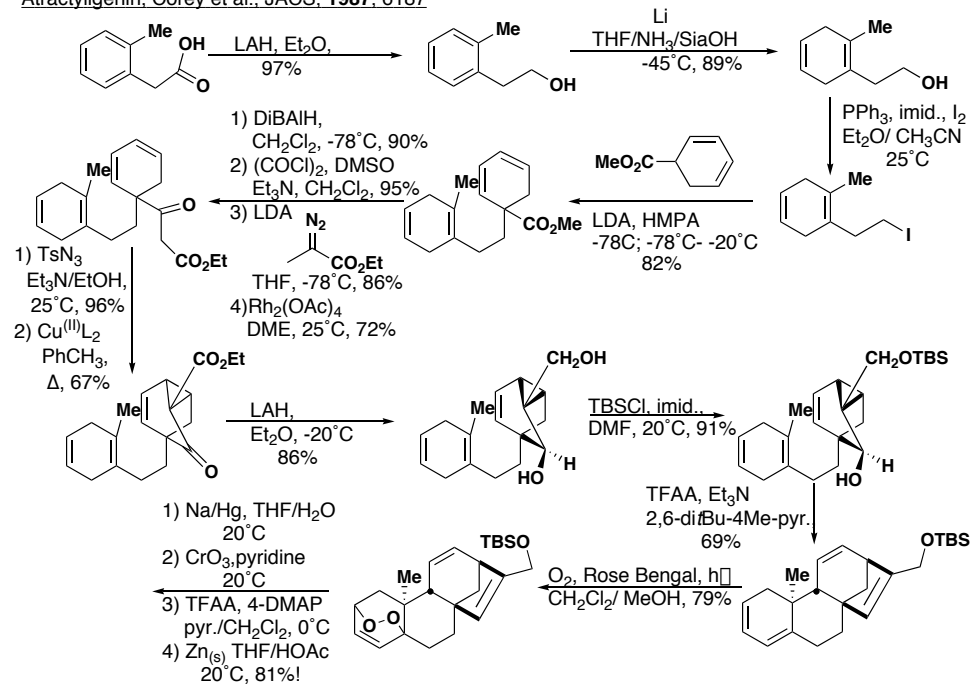
Horsfiline, Carriera et al., HCA 2000, 1175



Cafestol, Corey et al., JACS, 1987, 4717

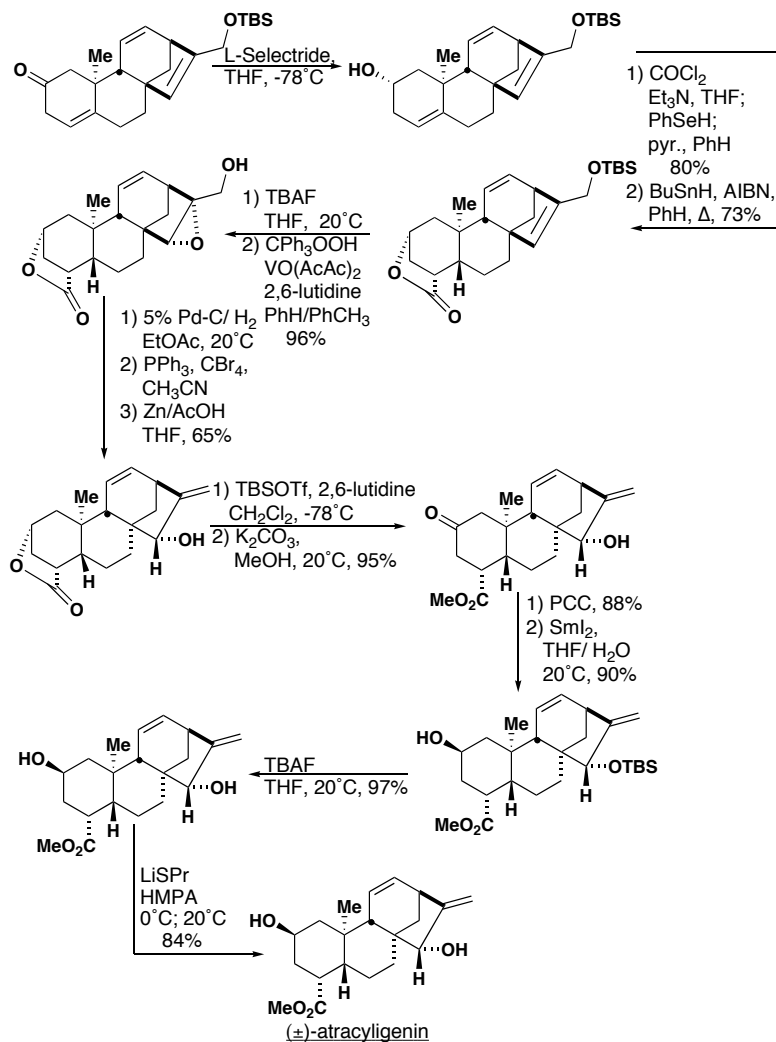


Atractyligenin, Corey et al., JACS, 1987, 6187



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Mycorrhizin, Dechloromycorrhizin, Smith et al., JACS, 1982, 2659

